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Introduction

Testosterone is the major circulating androgen in man.¹ Its deficiency is usually associated with adverse effects on body composition, bone density, sexual function, and mood and may also increase cardiovascular risk. The low molecular weight (M.W.=288) and hydrophobic nature ($\log P_{o/w} = 3.3$ and water solubility = 0.039 mg/ml at 37°C) in addition to first pass metabolism are favourable factors for transdermal delivery of testosterone. It is generally agreed that androgen replacement therapy should deliver physiological amounts (3-10 mg/day) of testosterone. This corresponds to targeted fluxes of 4-14 $\mu\text{g cm}^{-2} \text{hr}^{-1}$ over a reasonable surface area of 30 cm^2 . In this work, microemulsions of different compositions were considered to achieve these objectives. Single phase microemulsions are currently of interest to the pharmaceutical scientist as potential drug delivery vehicles due to their long term stability, ease of preparation, and considerable capacity for solubilization of a variety of drug molecules.

Results and Discussion

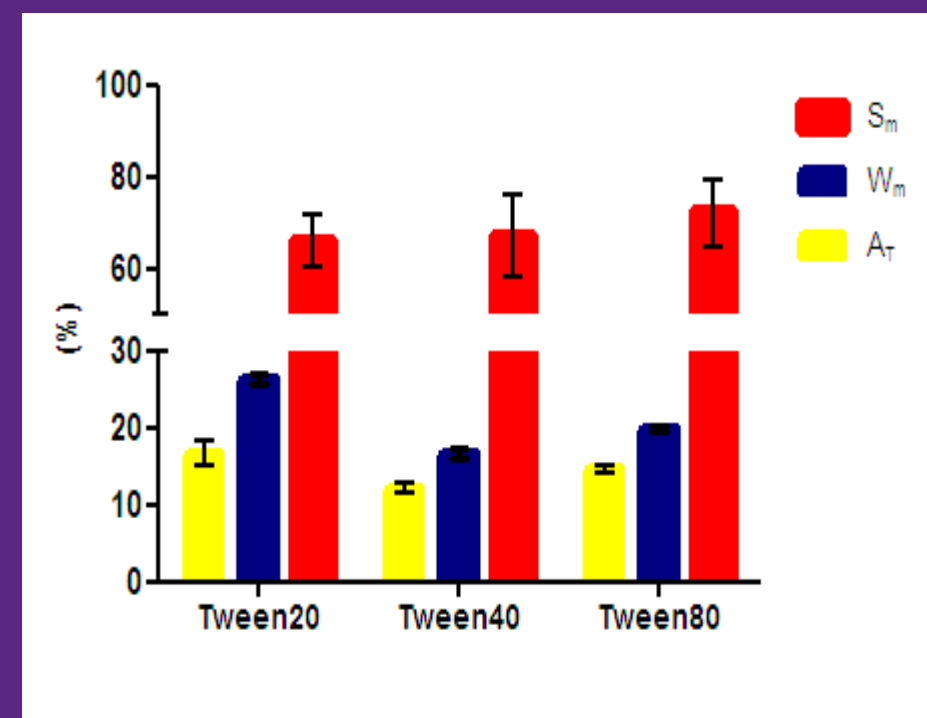


Fig.1 Effect of chain length on total monophasic area, maximum water of solubilization and minimum amount of surfactant needed for solubilization

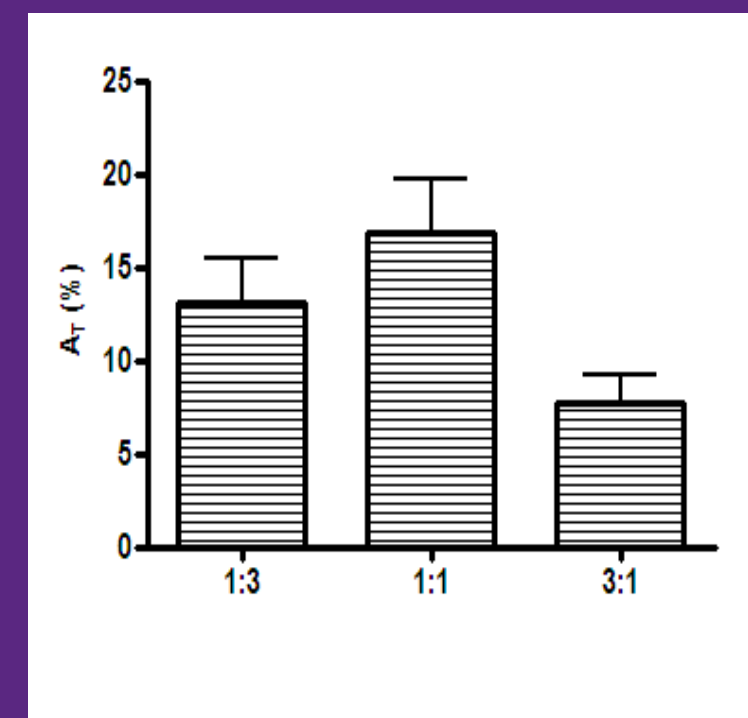


Fig.2 The optimum ratio between surfactant (Tween20) and cosurfactant (Transcutol)

Fig.3 Oleic acid / Tween20 / Transcutol / Water phase diagram where the mixing ratio of Tween20 : Transcutol is 1:1. 1Φ is the area of one phase region, $M\Phi$ is the area of multiple phase regions. W_m is the maximum amount of solubilized water, S_m is the amount of surfactant needed to obtain maximum water solubilization and $L20$ is a dilution line where the initial oil concentration is 20% (w/w).

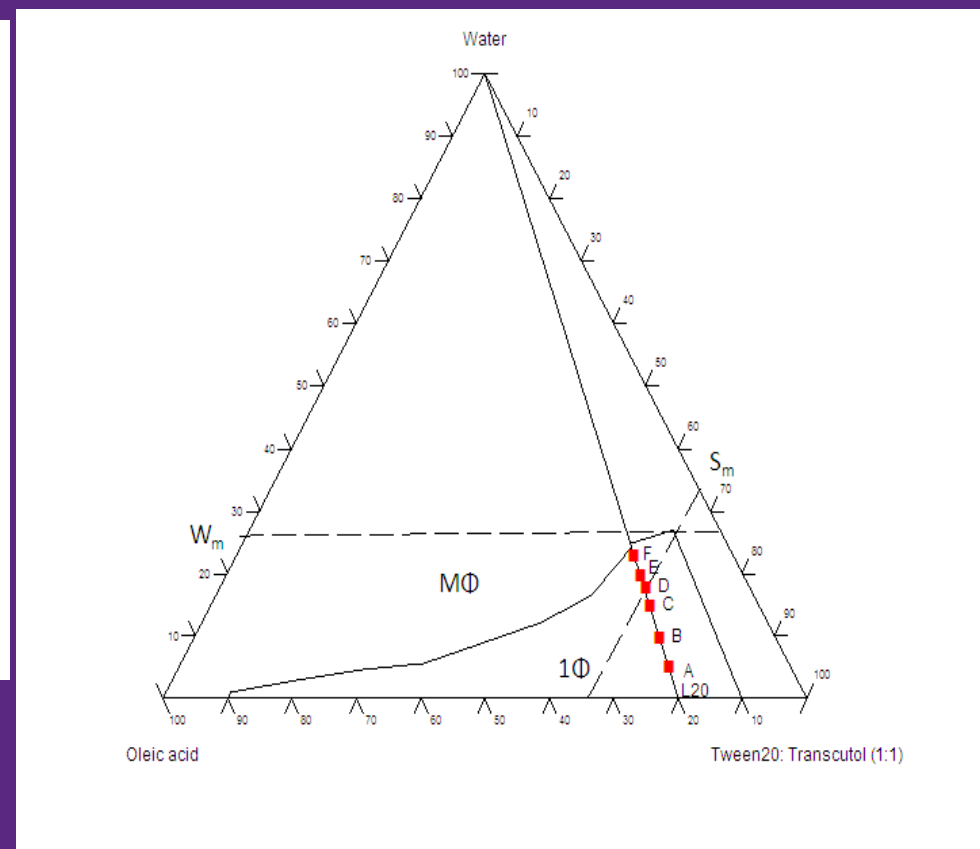


Table 3. Proton NMR chemical shifts of ME F different functional groups after incorporation of Drug (1% w/v) in ME F.

Functional group	δ (ppm)		$\Delta\delta$ ($\delta^1 - \delta^0$)
	In ME F (δ^0)	In ME F with 1% drug (δ^1)	
CH_3	0.855	0.888	0.031
CH_2	1.162	1.191	0.029
$\text{CH}_2\text{CH}_2\text{C}=\text{O}$	1.543	1.571	0.028
CH_2CH	1.989	2.02	0.031
CH_2CO	2.208	2.23	0.022
$\text{OCH}_2\text{CH}_2\text{O}$	3.646	3.668	0.022
CH_2OCO	4.171	4.190	0.019
OH	4.653	4.652	-0.001
CH	5.287	5.318	0.031

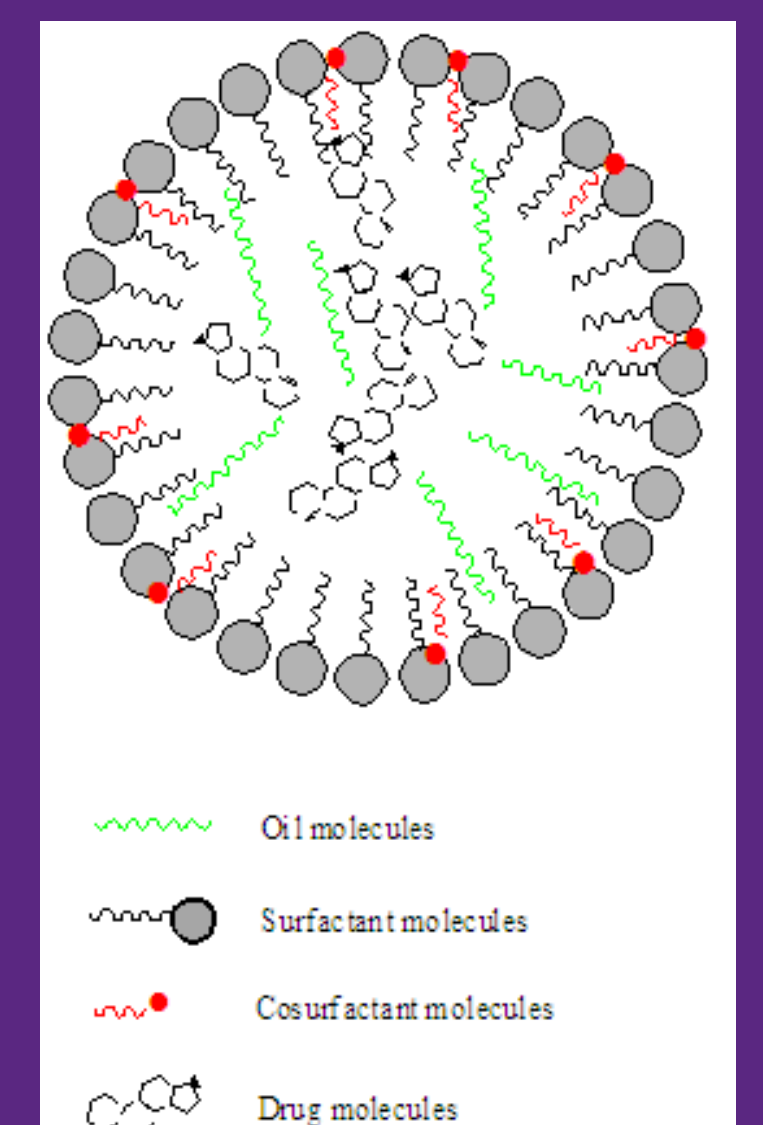
Table 4. Carbon NMR chemical shifts of ME F different functional groups after incorporation of Drug (1% w/v) in ME F.

Functional group	Oleic acid			Tween20			Transcutol					
	ME F (δ^0)	ME F with 1% drug (δ^1)	$\Delta\delta$ ($\delta^1 - \delta^0$)	ME F (δ^0)	ME F with 1% drug (δ^1)	$\Delta\delta$ ($\delta^1 - \delta^0$)	ME F (δ^0)	ME F with 1% drug (δ^1)	$\Delta\delta$ ($\delta^1 - \delta^0$)			
CH_3	14.117	14.806	0.689	CH_3	14.088	14.161	0.073	CH_3	14.843	14.916	0.073	
CH_2	22.896	22.926	0.030	CH_2	22.896	22.926	0.030	C-OH	60.863	60.914	0.051	
	29.477	29.513	0.036		29.477	29.513	0.036		C-O	70.067	70.126	0.059
	29.631	29.66	0.029		30.004	30.034	0.030			72.375	72.441	0.066
$\text{CH}_2\text{CH}_2\text{C}=\text{O}$	30.004	30.034	0.030	CH_2CO	25.131	25.183	0.052	CH_2CO	25.754	25.791	0.037	
CH_2CH	32.188	32.225	0.037	$\text{CH}_2\text{C}=\text{O}$	27.359	27.396	0.037	$\text{CH}_2\text{C}=\text{O}$	34.086	34.130	0.044	
CH_2CO	34.086	34.130	0.044	C-OH	60.863	60.914	0.051	C-O	70.067	70.126	0.059	
$\text{C}=\text{C}$	129.725	129.71	-0.015	$\text{C}=\text{O}$	175.782	175.73	-0.052	$\text{C}=\text{O}$	175.782	175.73	-0.052	
COOH	175.782	175.723	-0.059	$\text{C}=\text{O}$	175.782	175.73	-0.052					

Oleic acid functional groups are the most affected by testosterone presence. These results confirm the presence of the drug in the internal oily phase with small amounts at the surfactant/cosurfactant interface.

Where is testosterone in the microemulsion droplets?

Fig.6 Schematic diagram showing the position of testosterone and microemulsion components and their interactions.



Methodology

I- Construction of Phase diagram

II- Characterization of microemulsions by

- Particle size determination using a Malvern zetasizer
- Conductivity measurements
- pH
- Rheological measurements
- FTIR, and ¹H-NMR and ¹³C-NMR spectroscopy

III- Alteration of microemulsion microstructure after drug loading

IV- In vitro permeation of testosterone from selected microemulsions

V- Data Analysis

Results fitted to Fick's second law of diffusion to derive values of KH and D/H²

$$Q = \{KH\}C_{\text{veh}} \left[\frac{D}{H^2} t - \frac{1}{6} - \frac{2}{\pi^2} \sum_{n=1}^{\infty} \frac{(-1)^n}{n^2} \exp\left(-\frac{Dn^2\pi^2 t}{H^2}\right) \right]$$

Permeability coefficient was deduced from the relation:

$$K_p = (KH) \times \left(\frac{D}{H^2} \right) = \frac{K \cdot D}{H}$$

And the steady-state flux J_{ss} from:

$$J_{ss} = K_p \times C_{\text{veh}}$$

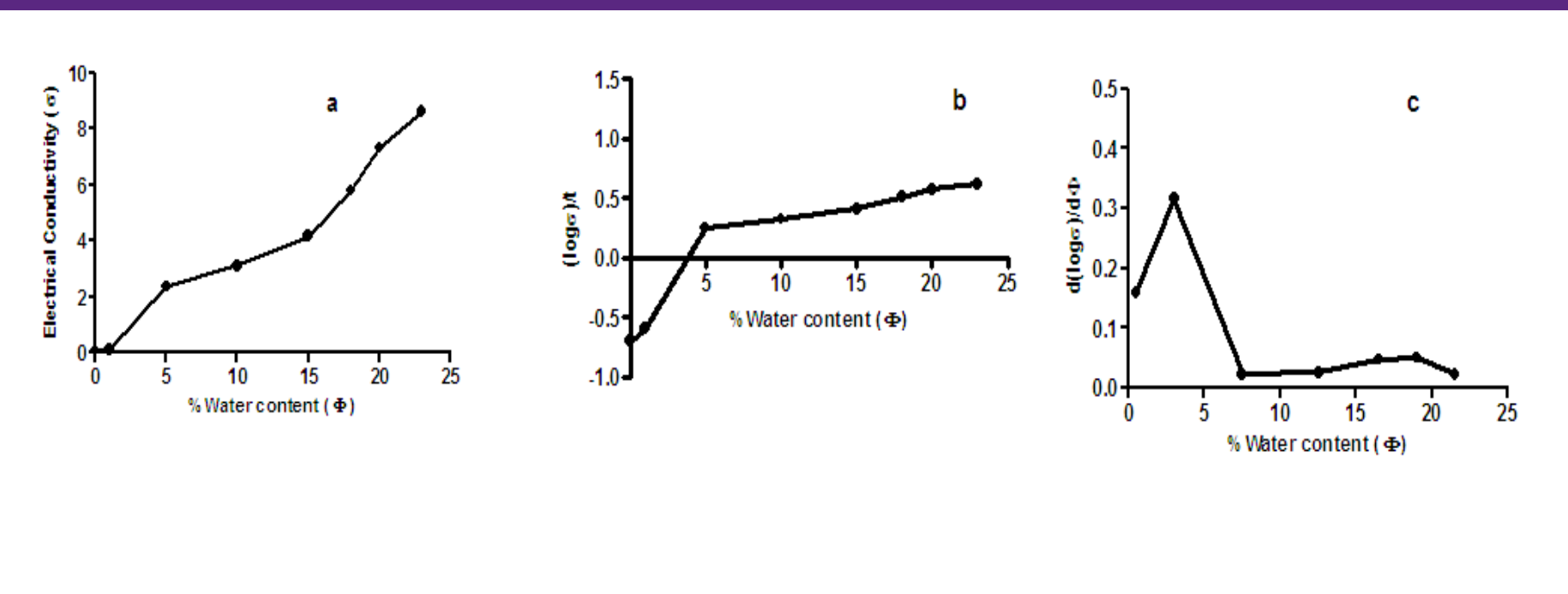


Fig.4. Plot of (a) electrical conductivity (σ), (b) $(\log \sigma) / t$ and (c) $d(\log \sigma) / d\Phi$ as a function of percentage water content (Φ) along the dilution line L20. A percolation threshold occurs at $\Phi=2.8$ where w/o microemulsion converts to o/w microemulsion.

Table 1 Selected microemulsion formulations (% w/w).

	A	B	C	D	E	F
Oleic acid	19	18	17	16.2	16	15.4
Tween20	38	36	34	32.9	32	30.8
Transcutol	38	36	34	32.9	32	30.8
Water	5	10	15	18	20	23

Table 2 Changes in the physical parameters of microemulsions after loading with drug (1% w/v).

ME	OH frequency (cm^{-1})		σ ($\mu\text{S/cm}$)		η (Pa.s)		pH	
	Unloaded	Loaded	Unloaded	Loaded	Unloaded	Loaded	Unloaded	Loaded
A	3449	3442	2.39 ± 0.16	2.35 ± 0.21	0.047 ± 0.011	0.036 ± 0.027	5.24 ± 0.01	5.30 ± 0.03
B	3436	3433	3.11 ± 0.27	3.63 ± 0.29	0.046 ± 0.012	0.049 ± 0.020	5.15 ± 0.04	5.14 ± 0.02
C	3434	3436	4.18 ± 0.64	5.04 ± 0.60	0.050 ± 0.016	0.062 ± 0.008	5.04 ± 0.03	4.97 ± 0.01
D	3426	3433	5.82 ± 0.53	6.35 ± 0.48	0.039 ± 0.018	0.062 ± 0.007	5.03 ± 0.03	4.97 ± 0.02
E	3431	3434	7.34 ± 0.23	7.20 ± 0.54	0.037 ± 0.014	0.040 ± 0.017	4.89 ± 0.02	4.90 ± 0.03
F	3401	3402	8.63 ± 0.59	8.71 ± 0.32	0.061 ± 0.007	0.073 ± 0.037	4.45 ± 0.03	4.45 ± 0.01

No statistical significant differences between loaded and unloaded microemulsions at $P < 0.05$.

According to these results, the drug is present in the oily phase.

Fig.7 Permeation of testosterone (1%) from the selected Oleic acid / Tween20 / Transcutol / Water microemulsions and fitting of the results to Fick's second law of diffusion.

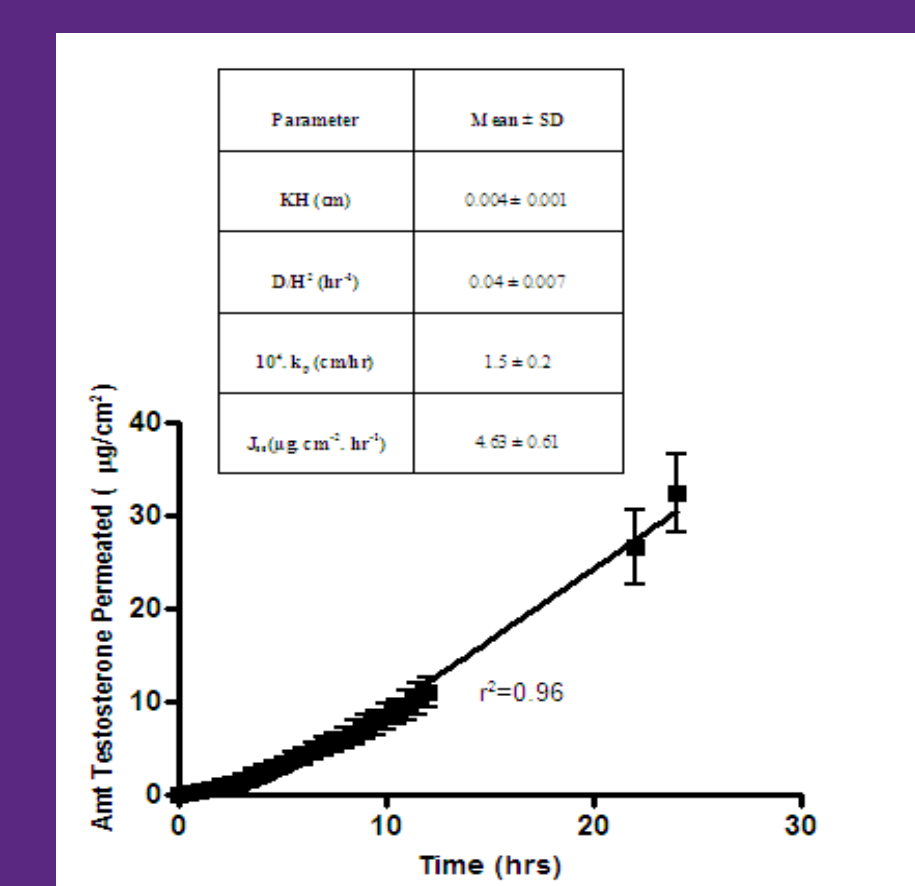
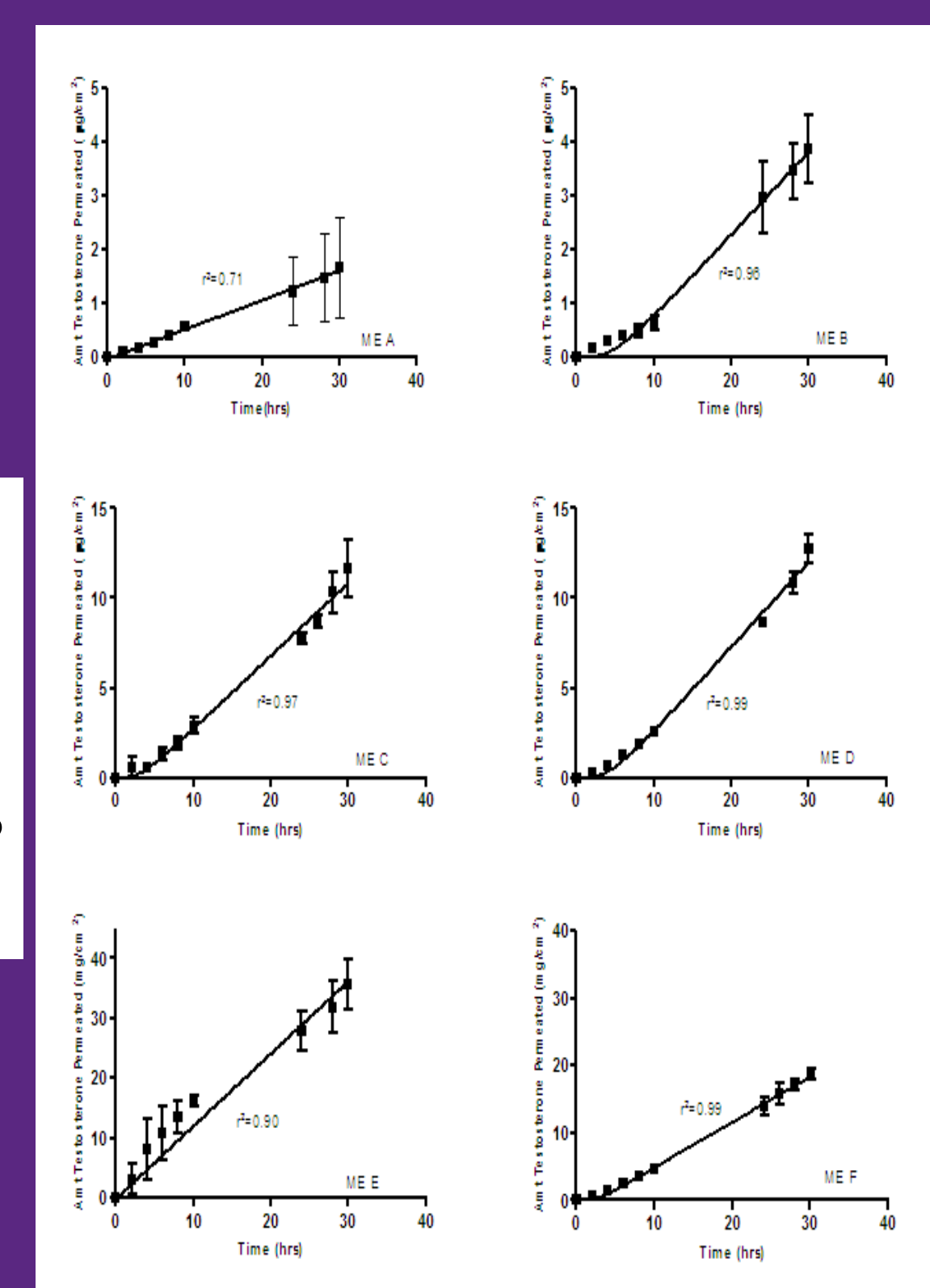
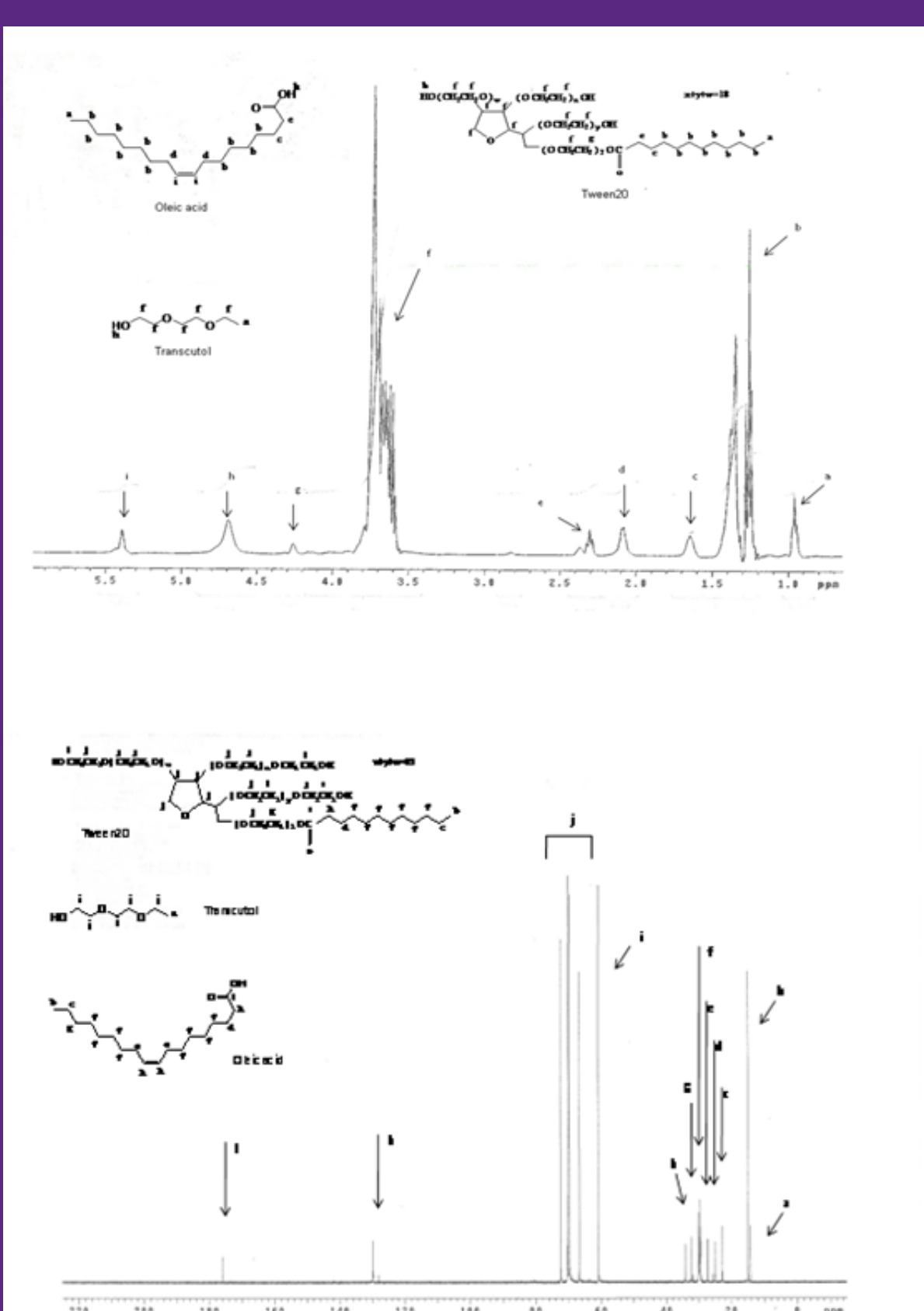


Fig.8 Permeation of testosterone (3%) from microemulsion E. The results were fitted to Fick's second law of diffusion.

Flux reached targeted values.

Fig.5 Proton NMR and Carbon NMR spectra of microemulsion F.



Conclusion

The microemulsion system studied offers a potential vehicle for the transdermal delivery of testosterone.

References

- Leichtnam ML, Rolland H, Wuthrich P, Guy RH. 2006. Identification of penetration enhancers for testosterone transdermal delivery from spray formulations. J Control Release 113:57-62.
- Leichtnam ML, Rolland H, Wuthrich P, Guy RH. 2006. Testosterone hormone replacement therapy: State-of-the-art and emerging technologies. Pharm Res 23:1117-1132.